



**B.Sc. II Semester (NEP) Degree Examination,
September/October - 2022
CHEMISTRY**

DSC - 2 : Models and Concepts in Chemistry

Time : 3 Hours

Maximum Marks : 60

Instructions : Answer *all* Sections.

SECTION - A

1. Answer the following sub-questions. Each sub-question carries **one** mark. **10x1=10**
- | | |
|--------------------------------------|---|
| (a) Define the covalent bond. | 1 |
| (b) What is electron gain enthalpy ? | 1 |
| (c) What are nucleophiles ? | 1 |
| (d) Write the Wurtz reaction. | 1 |
| (e) Define the unit cell. | 1 |
| (f) State Nernst Distribution law. | 1 |
| (g) What is Walden inversion ? | 1 |
| (h) What is Orientation effect ? | 1 |
| (i) What is precision ? | 1 |
| (j) What is absolute error ? | 1 |

SECTION - B

Answer **any four** of the following questions. Each question carries **five** marks.

4x5=20

2. Define electronegativity. Explain Mullikan-Jaffe's electronegativity scales. **5**
3. Explain with examples : **5**
- | | |
|--------------------------|--|
| (a) Addition reaction | |
| (b) Elimination reaction | |
4. Discuss the mechanism of SN^1 reaction with a suitable example. **5**
5. Explain powder-diffraction method. **5**
6. Derive Bragg's equation. **5**
7. Discuss on Determinate and Indeterminate errors. **5**



SECTION - C

Answer **any three** of the following questions. Each question carries **ten** marks.

3x10=30

8. (a) Explain the factors affecting ionization energy. **6**
(b) What is electron gain enthalpy ? Explain its trends in the periodic table. **4**
9. (a) Explain Free radical Substitution reaction of alkanes with a suitable example. **6**
(b) Explain types of bond breaking. **4**
10. (a) Explain mechanism of nitration reaction in Saturated carbon. **6**
(b) Discuss the factors affecting SN^1 and SN^2 reactions. **4**
11. (a) Explain classification of liquid crystals with an example. **6**
(b) Discuss principle and distribution law in Parker's process of desilverisation of lead. **4**
12. (a) Write a note on limit of detection [LOD] and limit of quantification [LOQ]. **6**
(b) Explain Absolute error and Relative error. **4**

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