

**M.Sc. II Semester Degree Examination, September/October - 2022****CHEMISTRY****21CHE2C7L : Electro-Quantum and Photochemistry**

Time : 3 Hours

Maximum Marks : 70

Note : Answer **any five** of the following questions with question no. **1** is **compulsory** and each question carries **equal** marks.

1. (a) Explain Debye-Huckel theory of strong electrolytes.
(b) Discuss in detail Helmholtz-Perin electrical double layer.
(c) Explain construction, working and applications of $H_2 - O_2$ fuel cell. **5+5+4=14**
2. (a) Explain the effect of temperature, current density and pH on overvoltage.
(b) Describe Buttler-Volmer equation
(c) Derive Tafel equation **5+5+4=14**
3. (a) Discuss the followings :
(i) De-broglie's equation
(ii) Heisenberg uncertainty principle
(b) Elucidate Schrodinger wave equation for a particle in one dimensional box.
(c) State and derive Variation theorem. **5+5+4=14**
4. (a) Explain Jablonski diagram of Photochemical reactions.
(b) Give laws of photochemistry.
(c) Write notes on :
(i) Photosynthesis
(ii) Quenching **5+5+4=14**
5. (a) Explain the kinetics of photochemical decomposition of CH_3CHO .
(b) Derive Stern-Volmer equation.
(c) What is Photocatalyst ? Explain the applications of TiO_2 photocatalyst. **5+5+4=14**
6. (a) Explain :
(i) Concentration overvoltage,
(ii) Hydrogen overvoltage
(b) Give postulates of quantum mechanics.
(c) Give applications of Schrodinger wave equation to rigid rotator. **5+5+4=14**



7. (a) Write notes on quantum yield and its determination.
(b) Discuss the working of Uranyl oxalate actinometer.
(c) Explain :
(i) Photosensitisation
(ii) Photodegradation **5+5+4=14**
8. (a) Explain reversible and irreversible electrodes with examples.
(b) What are operators ? Explain linear operators.
(c) Explain :
(i) Singlet and triplet states
(ii) Kinetics of photochemical combination of HCl **5+5+4=14**

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