

**21ICH1C3L****M.Sc. I Semester Degree Examinations, April/May - 2023****Industrial Chemistry****DSC 3 : Advanced Physical Chemistry**

Time : 3 Hours

Maximum Marks : 70

**Note :** Answer **any five** of the following questions with question No.1 (Q1) **compulsory**. Each question carries **equal** marks.

1. (a) Discuss Gibbs-Duhem equation and its applications. **4+3+3+4**  
(b) Give an account of Maxwell's relations of thermodynamics and their significance.  
(c) Explain the concept of fugacity and entropy.  
(d) Deduce Gibb's-Helmholtz equation.
2. (a) Explain the Lindeman theory of Unimolecular reactions. **5+5+4**  
(b) State and explain the activated complex theory.  
(c) Write a note on Hammett equations.
3. (a) Discuss the concept of liquid junction potential and its determination. **5+5+4**  
(b) Explain the effect of temperature, pressure and concentration on energetics of cell reaction.  
(c) Write a note on Debye-Huckel and Onsagar conductance equation of strong electrolytes.
4. (a) Give an account of BET equation and its significance. **5+5+4**  
(b) Write a note on kinetics of Michaelis-Menten equation.  
(c) Write briefly on Industrial applications of catalysts.
5. (a) Discuss the condensed phase rule for two component system with example. **5+5+4**  
(b) State the phase rule and what are the applications and limitations of phase rule ?  
(c) Explain Phase, component and degree of freedom.

6. (a) Write a short note on Branched chain reactions and its rate expression. **5+5+4**  
(b) Explain electrical double layer and its thermodynamics.  
(c) Discuss the physical significance of  $\kappa$  (Cuppa).
7. (a) Discuss the Protolytic and Prototropic mechanism in acid base catalysis. **5+5+4**  
(b) Explain the activation energies of catalyzed reactions.  
(c) Explain Nernst distribution law and its applications.
8. (a) Give an account on substitution effects on the rate of reactions. **5+5+4**  
(b) Discuss the Gouy-Chapman-Stern model of the electrical double layer.  
(c) Explain the factors ( $P^H$ , temperature and inhibitors) effecting the enzyme catalyzed reactions.

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