No. of Printed Pages : 1

21MNP3E2AL

Sl. No.

M.Tech. III Semester Degree Examination, April/May - 2024

MINERAL PROCESSING

Surface Chemistry

(NEP)

Time : 3 Hours

Maximum Marks: 70

Note : Answer **any five** of the following questions with question No. 1 (Q.1) is **Compulsory**. Each question carries **equal** marks.

1.	(a) (b)	Write a note on covalent solids. List out the properties of ionic compounds.	4 5
	(c)	What are colloidal compounds and how does they prepare ?	5
2.	(a)	Differentiate between the physiorption and chemisorption.	4
	(b)	Write a note on Fraundlich adsorption isotherm.	5
	(c)	Describe BET theory.	5
3.	(a)	Write a note on surface tension of solutions.	4
	(b)	Write a note on surface tension and its measurement.	5
	(c)	Explain surface tension and chemical composition.	5
4.	(a)	Define liquid-liquid interface and write the application of liquid-liquid interface.	4
	(b) (c)	Describe Stagnat layer at solid-liquid interface. Write a note on films at interfaces and emulsions.	5 5
5.	(a)	How static electricity and conductance affect on solid-gas interface.	4
	(b)	Write the effects of composition of liquid phases and electro-kinetic phenomenon.	5
	(c)	Describe electrical double layer and mutual repulsion.	5
6.	(a)	Explain polarity in covalent bond.	4
	(b)	List the postulates of Henry's law.	5
	(c)	What is the importance of thermodynamics of surface tension ?	5
7.	(a)	Write a note on importance of absorption and utilization.	5
	(b)	Applications of metallic compounds.	5
	(c)	Write a note on interfacial energy.	4
8.	(a)	Describe Polany's potential theory.	5
	(b)	Write a note on mechanical effects of adsorption.	4
	(c)	Describe multi-component system in liquid-liquid interface.	5

- o O o -

###